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of each kind in a chemical compound. Reveals the number of atoms of each element contained in a single molecule ...

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'Next' to see the next set of questions.

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CHAPTER 7 REVIEW Chemical Formulas and Chemical  
Compounds SECTION 3 SHORT ANSWER Answer the  
following questions in the space provided. 1. Label each of  
the following statements as True or False: True a. If the  
formula mass of one molecule is  $x$  amu, the molar mass is  $x$   
g/mol. False b. Samples of equal numbers of moles of two  
different chemicals

7 Chemical Formulas and Chemical Compounds  
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Quiz Modern Chemistry 33 . Name Section Quiz, continued  
Class Date 6. The Lewis structure of which of the following compounds does not follow the octet rule? a.  $\text{NH}_3$  b.  $\text{BF}_3$  c. d.  $\text{H}_2\text{O}$  7. In a double covalent bond, a. one atom has more than eight valence electrons.

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CHAPTER 5 REVIEW The Periodic Law SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. c In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's original design. (c) according to increasing atomic number. (d) based on when they were discovered. 2. d Mendeleev noticed that certain similarities in

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the ...

### 5 The Periodic Law

Modern Chemistry 137 Chapter Test Name Class Date

Chapter Test A, continued \_\_\_\_\_19. In the figure on the previous page, the pH at the equivalence point a. is equal to 7.0. b. is greater than 7.0. c. is less than 7.0. d. cannot be determined from the data given. \_\_\_\_\_20. In the figure on the previous page, the volume of titration standard

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