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Quantum Numbers • \_\_\_\_\_ • Used to \_\_\_\_\_ an \_\_\_\_\_ in an \_\_\_\_\_

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Note Taking Guide: Episode 304 Name Aditi Acharya Quantum Numbers  $n, l, m, s$  Used to describe an electron in an atom  $n$  Principle quantum number Represents main energy level of electron maximum # of electrons in an Energy level =  $2n^2$  Example: What is the maximum number of electrons that can be in the 5th main energy level? | The 2nd quantum number • Describes the orbital shape within an energy level • number of orbital shapes possible in energy level =  $n$  Orbital Shapes designated  $s, p, \dots$

~~note\_ep304aditi.pdf - orientation of orbital in space s ...~~

Note Taking Guide: Episode 304 Name \_ Quantum Numbers  $n, l, m, s$  Used to describe an electron in an

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atom \_ n principle quantum \_ number Represents main energy level of \_electron maximum # of \_electron \_ in an energy level \_ =  $2n^2$  \_ Example: What is the maximum number of electrons that can be in the 5th \_ main energy level? | The 2 nd quantum number \_\_\_\_\_ Describes the orbital \_\_\_\_\_ shape within an ...

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acceleration due to gravity = Example problems: Rhoda Tiller looks up to see a flowerpot fall off a ledge 9.2 m above her head.

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