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© OCR 2010 2 In the presence of acid,  $\text{H}^+(\text{aq})$ , aqueous bromate(V) ions,  $\text{BrO}_3^-(\text{aq})$ , react with aqueous bromide ions,  $\text{Br}^-(\text{aq})$ , to produce bromine,  $\text{Br}_2(\text{aq})$ . A student carried out an investigation into the kinetics of this reaction. (a) Balance the ionic equation for this reaction.  $\text{BrO}_3^- + \dots\dots\dots\text{Br}^- + \dots\dots\dots\text{H}^+ \dots\dots\dots\text{Br}_2 + \dots\dots\dots\text{H}_2\text{O}$

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