

## Ocr F322 June 2014 Answers

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### GCE Chemistry A

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F322/01 Mark Scheme June 2013 6 Question Answer Marks Guidance 1 (a) (iii) B has less surface (area of) contact OR OR A AND B has fewer van der Waals ' forces OR B has weaker van der Waals ' forces OR OR A So less energy needed to break the intermolecular forces in B OR OR A 2 Both answers need to be comparisons

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GCE Chemistry A

OCR LEVEL 1/2 CABRGE ATOALS CREATVE iEA 3 INTRODUCTION This resource brings together the questions from the June 2014 examined unit (R081), the marking guidance, the examiner's comments and the exemplar answers into one place for easy reference.

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All examiners are instructed that alternative correct answers and unexpected approaches in candidates' scripts must be given marks that fairly reflect the relevant knowledge and skills demonstrated. Mark schemes should be read in conjunction with the published question papers and the report on the examination.

GCE Chemistry A

F324 Mark Scheme June 2014 10 Question Answer Mark Guidance 2 (a) FIRST react all with Tollens' reagent AND silver mirror/ppt/solid (formed) with compound D OR with Fehling's/Benedict's solutions AND (brick-red/orange) solid/precipitate (formed) with compound D ALLOW NOTE: eliminates D + [O] THEN react C and E with H<sub>2</sub>SO<sub>4</sub> / H

GCE Chemistry A

F325 Mark Scheme June 2014 3 Question Answer Marks Guidance 1 (a) (i) 2K+(g) + S<sub>2</sub>(g) + e<sup>-</sup> → 2K+(g) + S<sub>2</sub><sup>2-</sup>(g) 2K(g) + S(g) – – 3 Mark each marking point independently Correct species AND state symbols required for each mark For S<sub>2</sub><sup>2-</sup>, DO NOT ALLOW – S<sub>2</sub> For e<sup>-</sup>, ALLOW e For e only, IGNORE any state symbols added ALLOW k and s It can be very difficult distinguishing K from k; S from s

GCE Chemistry A

CHEMISTRY A F322 Chains, Energy and Resources \*OCE/T73768\* ... Candidates answer on the question paper OCR Supplied Materials: • Data Sheet for Chemistry A (Inserted) Other Materials Required: • Scientific calculator Wednesday 3 June 2009 Morning Duration: 1 hour 45 minutes THIS IS A NEW SPECIFICATION FOR EXAMINER'S USE

ADVANCED SUBSIDIARY GCE CHEMISTRY A F322

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