

Physical Science Final Study Guide

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Physical Science Final Exam Study Guide I. Chemistry 1. What is the difference between a homogeneous mixture and a heterogeneous mixture? 2. List 2 examples of a homogeneous mixture and 2 examples of heterogenous mixtures. 3. List 2 examples of a pure substance and 2 examples of a mixture. 4.

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Physical Science Final Study Guide. STUDY. PLAY. Kinetic theory. a theory of the behavior of gases that assumes heat is a process of energy transfer and the internal energy of a gas is the total energy of its particles. boiling point. the temperature at which a heated liquid turns into gas.

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Physical Science Final Study Guide. STUDY. PLAY. There are many different ways to classify matter, but one way is to use only two categories for ALL matter. All matter can be classified as either: Mixtures or substances. A substance made of two or more elements that cannot be separated by physical means is called a:

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Physical Science Final Exam Study Guide. STUDY. Flashcards. Learn. Write. Spell. Test. PLAY. Match. Gravity. Created by. dsomerville. Combo with 9-1 Describing and Measuring Motion and 27 others. Terms in this set (264) motion. the state in which one object's distance from another is changing. reference point.

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8th Grade Physical Science Final EXAM Review Packet 1. Match the following vocabulary words with the correct definition on the right ____ Physical Science ____ Period (of Pendulum) ... The study of energy and matter F. Testable idea based on background knowledge G. The application of scientific knowledge

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Physical Science - 1st Semester - Final Exam Study Guide - (2015-2016) 1. Physical Science Basics a. What is the formula for calculating density? ____ i. What is the density of a cube that is 15 g and has a volume of 3 cm 3? ____ ii. Compare the density of solid H 2 O to liquid H 2 O, and explain what observable effect this has.

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Physical Science Final Study Guide: Definition. All answers are True: Term. Definition. Metals contain metallic bonds: Term. Definition. Alloys are mixtures of different metals: Term. Definition. In a chemical reaction the mass before = the mass after: Term. Definition. C + O2 ---- CO2 is a chemical equation:

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Physical Science Final Study Guide 2nd Semester. STUDY. PLAY. Chemical Property. any characteristic of a substance that allows a substance to undergo a change that results in a new substance. To observe some chemical properties, the original substance being observed must ____ into one or more different substances.

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Health Science 430 Final Exam Study Guide: Nutrition and Physical Activity 1. Obesity disparities 2. Difference between overweight and obesity 3. What each food group provides the body 4. Anorexia and bulimia 1. Obese disparities: higher risk for blacks, hispanics and low socioeconomis status-Prevalence of obese adults have nearly tripled in past 20 years-More than 30% of adults obese now ...

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