

## Plus Two Chemistry Chapter Full Notes

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Kerala Plus Two Chemistry Chapter Wise Questions and Answers Chapter 11 Alcohols, Phenols and Ethers. Very Short Answer Type Questions (Score 1) Question 1. Glycerol is \_\_\_\_\_. Answer: 2, 3, 1 dihydroxy propanal. Question 2. Picric acid is a. 2, 4, 6 – trinitrotoluene

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Kerala Plus Two Chemistry Notes Chapter 3 Electrochemistry. Electrochemistry. It is the study of production of electricity from I energy which is released during spontaneous chemical reactions, also the use of electrical energy to bring about non-spontaneous chemical trasformations. Electrolyte.

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Plus Two Chemistry Chapter Full Notes Kerala Plus Two Chemistry Notes Chapter 3 Electrochemistry. Electrochemistry It is the study of production of electricity from I energy which is released during spontaneous chemical reactions, also the use of electrical energy to bring about non-spontaneous chemical trasformations.

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$2\text{Fe}(s) \rightarrow 2\text{Fe}^{2+} + 4\text{e}^{-}$   $E^{\circ} = -0.44\text{ V}$ . Electrons released at anodic spot move through metal and go to another spot on the metal and reduce  $\text{O}_2$  in presence of  $\text{H}^{+}$ . This spot behaves as cathode.  $\text{O}_2(g) + 4\text{H}^{+}(aq) + 4\text{e}^{-} \rightarrow 2\text{H}_2\text{O}(l)$   $E^{\circ} = 1.23\text{ V}$ . The overall reaction is,  $2\text{Fe}(s) + \text{O}_2(g) + 4\text{H}^{+}(aq) \rightarrow 2\text{Fe}^{2+} + 2\text{H}_2\text{O}(l)$   $E^{\circ} = 1,67\text{V}$

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Kerala Plus Two Chemistry Notes Chapter 2 Solutions. Solution A homogeneous mixture of two or more sub-stances is knows as solution. Solute : The substance present in smaller amount in a solution is called solute. Solvent: The substance present in larger amount in a solution is called solvent.

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Kerala Plus Two Chemistry Notes Chapter 9 Coordination Compounds. Coordination chemistry – branch of chemistry which deal with the complex compounds formed by transition and other metals. Chlorophyll, haemoglobin and vitamin B 12 are coordination compounds of Mg, Fe and Co respectively. Werner ' s Theory of Coordination Compounds The main postulates are,

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$2\text{Ca}(\text{OH})_2 + 2\text{Cl}_2 \rightarrow \text{Ca}(\text{OCl})_2 + \text{CaCl}_2 + 2\text{H}_2\text{O}$   $\text{Cl}_2$  is a powerful bleaching agent.  $\text{Cl}_2 + \text{H}_2\text{O} \rightarrow 2\text{HCl} + [\text{O}]$  Coloured substance +  $[\text{O}]$  colourless substance. Uses: For bleaching wood pulp, cotton and textiles; for the preparation of insecticide, pesticides and other organic solvents, e.g.  $\text{CHCl}_3$ , DDT, BHC etc. Hydrogen Chloride (HCl): Preparation:

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