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2 Chemistry, First Canadian Edition 1.17 An organic compound consists of molecules made from carbon, hydrogen and oxygen atoms. There can be other atoms - for example, nitrogen, sulfur. 1.19 1. Acetylsalicylic acid, the common mild pain reliever, is a derivative of the natural product, salicin, obtained from willow bark. 2.

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4-5) 2. (b) 4.2 g Chapter 1 Section 1.4  
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<http://www.science.nelson.com/0176259864/studquizzes.html>. Unit 2 Review, pages 272-279.  
is  $4n - 2$  electrons, where  $n$  is the principal quantum number. 82. (a) The element with the electron configuration  
 $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6$  is krypton, since



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the atomic number of krypton is 36.

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waste in a chemistry laboratory You can  
review prerequisite concepts and skills on  
the Nelson Web site and in the Appendices. A  
Unit Pre-Test is also available online.

Prerequisites Table 1 Mass and Volume of a  
Solid Mass (g) Volume (mL) 1.2 3.6 1.8 5.5  
2.3 6.9 3.1 9.2 6.9 20.7

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Indicate whether the sentence or statement is  
true or false. 1. Nuclear changes generally  
absorb more energy than chemical changes. 2.  
In exothermic reactions, the reactants have  
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True/False. Indicate whether the sentence or  
statement is true or false. T F. 1. The  
region in space where an electron is most  
likely to be found is called an energy level.  
T F. 2.

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gases combine to form oxygen and nitrogen dioxide gases. The relative numbers of molecules of the four respective species are 1:1:1:1 - i.e. the stoichiometric coefficients. 2.8 The atom is

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