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An electron is contained in a three-dimensional rectangular box given by $0 \leq x \leq a$, $0 \leq y \leq b$, and $0 \leq z \leq c$. The solutions of the Schrödinger equation are specified by the quantum numbers n_x , n_y , and n_z . Recall that the normalized wave function is...with energy...where n_x , n_y , and n_z are positive integers.

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r , and $l(l+1) = m^2 - 1/4$, which is to say, $l^2 + l + 1/4 = m^2$, or $(l + 1/2)^2 = m^2$, or $l = |m| - 1/2$. [Our present equation depends only on m^2 , and hence is the same for either sign, but the solution to Problem 4.39 assumed $l + 1/2 > 0$ (else u is not normalizable), so we need $|m|$ here.]

Quoting 4.39: $E = \hbar^2 k^2 / 2m$.

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