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Standardized Test Practice

World of Chemistry 28 Copyright © Houghton Mifflin Company. All rights reserved. Name _____ Section _____ Date _____ 10. A student drew the following electron box ...

Chapter 11 Standardized Test Practice

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temperature she tested.

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answer for each question is indicated by a . 1: Matter is composed of tiny particles called _____.
Need a Hint? A) oxygen: B) atoms ...

Standardized Test Practice-English

Answers by Chapter. Chapter 1 - Introduction to Chemistry. Chapter 2 - Matter and Change.
Chapter 3 - Scientific Measurement. Chapter 4 - Atomic Structure. Chapter 5 - Electrons in
Atoms. Chapter 6 - The Periodic Table. Chapter 7 - Ionic and Metallic Bonding. Chapter 8 -
Covalent Bonding.

Chapter 4 - Atomic Structure - Standardized Test Prep ...

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Answers

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Chapter 6 Standardized Test Practice 1. The actual number of atoms of each element contained in a molecule is given by the molecule's: A. empirical formula. C. mass percent formula. B. molecular formula. D. percent composition. 2. What is the molar mass of the sodium sulfate Na_2SO_4 ? F. 71.06 g/mol G. 94.05 g/mol H. 142.05 g/mol J. 215.13 g/mol 3.

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