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### ~~Straighterline Chemistry Answer Key~~

Straighterline Chemistry Answer Key straighterline chemistry answer key The initial volume of a system containing 1 mole of an ideal gas at  $292^{\circ}$  Kelvin and 3 atm is 8 L. If the gas is cooled at constant volume until the pressure falls to 1.2 atm, then the gas is heated and expanded at constant pressure (1.2 atm) until the volume is 20 L and ...

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Answer: 4.5oC. Measuring Heat of Reaction for a simple system. An important thermodynamic property in a reaction is the molar enthalpy of reaction ( $q_{\text{sys}}$ ). The molar enthalpy of reaction is defined as the heat absorbed by the system from the surroundings in a reaction when one mole of product is formed.

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