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The vibrating molecules emitting the radiation could have only certain discrete amounts of energy, given by

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People $E_n = nhf$, where n is called a quantum number, F is the frequency, and h is Planck's constant given by $h = 6.626 \times 10^{-34}$ joule?seconds.

Molecules emit energy in units called quanta, now called photons.

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$$\int_V |\psi(r,t)|^2 dr = 0. \quad (6)$$

This result is proven to be in agreement with the Schrödinger equation, and it then becomes natural to

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de?ne the probability
density ρ and probability
current density j according
to $\rho(r, t) = |\psi(r, t)|^2$, (7)
 $j(r, t) = \text{Re}.$

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Quantum mechanics is the branch of physics dealing with physical phenomena at microscopic scales, particularly with notions of probability. Probability in Matter. Considering light as particles (photons), the

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Probability per volume of finding a photon in a given region of space at a given time is proportional to the number N of photons per unit volume at that time and to the intensity:

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particularly the mathematics
used, or think that it is
irrelevant to chemistry and
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apply to the real world and,
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Special Relativity and
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Guide Lesson 1: Relativity
Relative Velocity Velocity
is relative to an observer's
frame of reference. Consider

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2 observers: one in boat and one on shore watching boat. Observer in boat would look at water and notice how fast he is moving relative to it. Observer on shore would look at water and boat and notice how they appear faster.

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Relative velocity of object can be calculated by adding up all the separate velocity vectors.

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