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2 Answer TEACHING

TRANSPARENCY WORKSHEET The

Activity Series Use with

Chapter 9, Section 9.2 1.

For each of the following
pairs of elements, underline
the one that would replace
the other element in a

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WORKSHEET Ionic Bonds Use
with Chapter 7, Section 7.2

1. How many valence
electrons does a neutral

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magnesium (Mg) atom have? 2. What is the charge on a magnesium ion? What does magnesium have to do to form such an ion, and why does it tend to do so? 3. How many valence electrons does a single neutral chlorine (Cl)

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make great conversation, but
if you want to improve your

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makes a lot of sense.

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6. Considering your answer
to question 5, how many

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electrons can fit in the entire 5th energy level? can (Ave1 is a ha AC5a) 50đ 7. Write the notation for an electron spinning clockwise in a p sublevel in the 4th energy level.

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WORKSHEET Atomic Orbitals 1.

What is the shape of an s
orbital? Class 16 Use with
Chapter 5, Section 5.2 2.

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What is the relationship between the size of an s orbital and the principal energy level in which it is found? 3. What is the shape of a p orbital? How many p orbitals are there in a ublevel? 4.

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Teaching Transparency 18-1

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Connecting Math to Physics

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