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Periodic Table 1. Identify the number of valence
electrons in each of the following elements. a. Ne d.

Mg f. Cl h. Si 2. Identify the energy level of the

valence electrons in each of the following elements. a.
c. d. e. H Ar 1 3.

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Teaching Transparency The Periodic Table Setting the
Table. One way to teach the periodic table is to make

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it into a table-setting exercise; this is perhaps best for children ages 8 to about 10.

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How are nonmetals designated in this periodic table?
__boxes are tinted yellow___ 16. What can be said
about the electron configurations of all the elements
in a group? _their valence electron configurations are
identical___ The s-, p-, d-, and f-Block Elements . 1.
What are the four sections, or blocks, of the periodic
table?

The Periodic Table

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On the periodic table below, identify the s-block, p-block, d-block and f-block elements. Then make the connection between the various orbitals and the elements in the periodic table. Use the noble gases to write abbreviated electron configurations for the following: d. Mg s² 3d¹⁰ 4s² j. l- a 1-1 k. Ag

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