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Note: Not all of the reactions will occur. For those that do not, write no reaction.

1. A piece of copper is dropped into a container of water. No reaction
2. Liquid bromine is added to a container of sodium iodide crystals.
 $\text{Br (l)} + 2\text{NaI(s)} \rightarrow 2\text{NaBr(s)} + \text{I}_2\text{(s)}$
3. An aluminum strip is immersed in a solution of silver nitrate.

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The Ultimate Chemical Equations Handbook SnCO tin(II) carbonate sodium hydrogen carbonate NaHCO₃ 1. 2. 3. 4. 5. 6. 8. 9. 10. 11. vanadium(V) oxide H₂O dihydrogen monoxide (NH)₄ CO ammonium oxalate 4 224 polonium(VI) thiocyanate Po(SCN)₆ tetraphosphorus decaoxide P₄O₁₀ 12. 13. 15. 16. 17. 18. 19. manganese(VII) oxide copper(II) dihydrogen phosphate

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8: ROUND 3 Exercise 8-1: Using the activity series, predict and balance the following single replacement reactions.

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Americium—named after the Americas (after all, one must be patriotic!).

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question requiring the determination of

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