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## Chemical Bonds

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A chemical bond is  
formed when electrons  
are shared between two  
atoms. There are three  
types of bonds: covalent  
bonds, polar covalent

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bonds and ionic bonds.

The simplest example of bonding is demonstrated by the  $H_2$  molecule. We

can see that each

hydrogen atom has a single electron from the periodic table.

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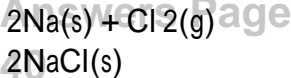
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## 7.2 IONIC BONDING

The chemical bond formed by transfer of electron from a metal to a non- metal is known as ionic or electrovalent bond . For example, when sodium metal and chlorine gas are brought into contact, they react

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violently and we obtain sodium chloride. This reaction is shown below:



## 7 CHEMICAL BONDING

There are many types of chemical bonds that can form, however the 3 main types are: ionic, covalent, and metallic bonds. You must

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become familiar with how they work and the differences between the 3 types. I. Ionic bonding:

Model 1 is a description of what chemists call ionic bonding. Ionic bonding occurs strictly between metal and nonmetal atoms.

The Different Types of  
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Accordingly, there are  
different types of  
chemical bonds, like,  
Ionic or electrovalent  
bond Covalent bond Co-  
ordinate covalent bond  
In addition to these we  
have a special kind of

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bond called hydrogen bond. Let us discuss about different types of bonds, their formation and the properties of the compounds so formed.

## 5.2 Ionic or Electrovalent Bond

### CHEMICAL BONDING

There are two main types and some secondary types of chemical bonds:

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1 Ionic bond Ionic bonding involves a transfer of an electron, so one atom gains an electron while one atom loses an electron. One of the resulting ions carries a negative charge (anion), and the other ion carries a positive charge (cation).

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gives an electron to a  
chlorine atom Force that  
causes a ...

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